

Techno-economical and Environmental Study of Utilizing Alternative Fuel and Waste Heat Reuse in a Cement Plant

Emad Benhelal¹; Reza Azin^{2,3,*}; S. Mostafa Jafari Raad²

¹Faculty of Chemical Engineering, University Technology Malaysia (UTM), Skudai, Johor Bahru, Malaysia

²Chemical Engineering Department, Faculty of Engineering, Persian Gulf University (PGU), Bushehr, Iran

³Persian Gulf Science and Technology Park, Bushehr, Iran

*Corresponding author.

Email: reza.azin@pgu.ac.ir

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Abstracts

In this work, a cement plant was simulated to estimate heat losses and pollution emission in the process. Also, a waste heat recovery system was introduced to use two main sources of heat losses, i.e. flue gas and hot vent air streams and produce steam and power. Moreover, the use of natural gas as an alternative to the current energy source, fuel oil, was studied in two cases, associated with and without waste heat recovery system. Results showed

that 34.28% of initial energy was lost in the base case, 48% of which is from flue gas and hot vent air streams. Also, changing the fuel source from fuel oil to natural gas results in CO₂ emission rate to decrease from 118,693 to 115,367 kg/hr, and emission of NO₂ and SO₂ was reduced to nearly 100%. In addition to environmental benefits, economical analyses suggest the use of waste heat recovery system as well as change of fuel for this plant.

Key words: Cement plant; Waste heat recovery; Alternative fuel; HYSYS simulation; Air pollution reduction

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NOMENCLATURE

GHG	Greenhouse Gases
UNFCCC	United Nations Framework Convention on Climate Change
ppm	Part per million
Tg CO ₂ Eq	Teragrams of CO ₂ Equivalent
WHRSG	Waste Heat Recovery Steam Generator
VOC	Volatile Organic Compound
TDF	Tyre-Derived Fuel
XRD	X-Ray Diffraction
XRF	X-Ray Fluorescence
PAHs	Polycyclic Aromatic Hydrocarbons
GA	Genetic Algorithm
ORC	Organic Rankine Cycle
PCDD/Fs	Poly Chlorinated Dibenzo-p-Dioxins and Dibenzo Furans
β-C ₂ S	Beta-Dicalcium Silicate
C ₃ S	Tricalcium Silicate
C ₃ A	Tricalcium Aluminate
C ₄ AF	Tetracalcium Alumino ferrite
HHV	Higher Heating Value
MW	Molecular Weight
PFD	Process Flow Diagram
L.O.I	Loss of Ignition

INTRODUCTION

Since two past decades and due to dramatic increase of environmental threats, societies have pressured their governments to discuss environmental issues, and the accumulation of greenhouse gases (GHG) emissions in the atmosphere is one of these controversial debates. The Kyoto Protocol and Copenhagen Conference are two agreements made under the United Nations Framework Convention on Climate Change (UNFCCC). Under these agreements some countries committed to reduce their Greenhouse Gas (GHG) emissions to the certain levels.

Carbon dioxide as the most important and abundant gas among greenhouse gases is generated and emitted by human related activities. The concentration of this gas in the atmosphere during preindustrial time was about 290 part per million (ppm). Since the beginning of the industrial revolution and due to dramatic increase of industrial activities as well as public and individual transportation this concentration was increased and today has reached to 360 ppm and is being increased by the rate of 0.3-0.4 % per year (Nazmul, 2005).

Industrial activities due to combusting huge quantity of fossil fuels in order to generate required energy for their process are the most important factors of carbon dioxide emission in the world. In addition to carbon dioxide generated via fuel combustion, in some processes such as production of ammonia, petrochemicals, titanium dioxide and cement plenty of carbon dioxide is generated by the inherent chemical reaction during the process.

Cement production is one of the most energy intensive and carbon dioxide emitter industrial processes in the world. It contributes approximately five percent of the total CO₂ emitted worldwide, emitting nearly 810 kg of CO₂ for every 1000 kg of cement produced (Hendriks et al., 1999).

In view to the latest ranking of the most carbon dioxide emitter industrial processes in U.S, cement industry has been ranked in the second place which has emitted almost 41.1 Tg CO₂ Eq (teragrams of CO₂ equivalent) in 2008 (U.S.A, Environmental Protection Agency, 2010). This huge quantity of emission is generated through 4 major sources, including fossil fuel combustion (about 40% of the industry's emissions), transport of raw materials (about 5%), and generation of electricity consumed by the cement manufacturing operations (about 5%). The remaining cement-related emissions (about 50%) originate from the process that converts limestone to carbonate oxide (Nazmul, 2005).

In addition to intensive fuel consumption and inherent reaction, the presence of some problems in conventional plant can also indirectly increase the quantity of carbon dioxide emission in the process. These problems which can increase fuel consumption and result in CO₂ emission are classified as follow:

- losses of large quantity of thermal heat through flue

gas and hot air streams

- heat losses through cooler stack, kiln, calciner, cyclones and ducts' shell
- utilizing high carbon content fuels as the major source of thermal energy

Therefore and in order to solve these problems and mitigate energy requirement as well as carbon dioxide emission in cement production process, it is necessary to explore, find and implement proper methods such as process optimization, heat losses reduction and using alternative fuels.

Since past decade enormous worthwhile studies have been conducted on energy saving and alternative fuel utilization in cement plant aimed to operate the process energy efficiently and environmental friendly. For example, Khurana et al. (2002) investigated heat recovery from two main sources of thermal heat losses, i.e. flue gas and hot air streams and producing steam using Waste Heat Recovery Steam Generator (WHRS). Implementing WHRS indicated that it was possible to save up to 10% of initial energy consumption in the process. Prisciandaro et al. (2003) analyzed the experimental results of the emission of alternative fuels replaced with conventional fuel in two different cement plants in Italy using Statistical Student's t-tests, stepwise linear regression models and factor analysis. Results indicated that if less than 20% of regular fuel is replaced with tyre, stack emissions (mainly NO_x, SO₂ and C₀) were slightly increased, but remaining almost always below the law imposed limits. In contrast, if waste oils were used polluted gas emissions were significantly decreased. Mokrzycki et al. (2003) studied ecological and economical features of utilizing alternative fuels made from waste, called PASi and PASr in the Lafarge Cement Polska S.A. group in Poland. They showed that combusting these fuels would result on permissible values of emissions. The results also confirmed that utilizing these alternative fuels is both environmentally-friendly and profitable to cement industry and society. Gabel & Tillman (2005) simulated nine scenarios of using recovered materials and alternative fuels. Simulations results showed that emissions of CO₂, NO_x, SO₂, CO, VOC, CH₄ and dust could be mitigated up to 80% due to use of recovered material and alternative fuel. Rasul et al. (2005) used data obtained from industry in Indonesia to estimate heat losses through the kiln system and Cooler. Furthermore, he identified thermal energy conservation opportunities which showed that about 1.264 × 10⁵ US dollars per year could be saved by replacing industrial diesel oil with waste heat recovery from kiln and cooler exhaust for drying of raw meal and fuel and preheating of combustion air. Engin & Ari (2005) found that 40% of initial energy was lost through hot streams (19.15%), cooler stack (5.61%) and the kiln's shell (15.11%) in a cement plant in turkey. In order to reduce heat losses and run the process in optimized form, they utilized (WHRS) to produce steam by hot air stream, hot

flue gas to preheat raw material and also secondary shell as insulation layer for the kiln to decrease thermal energy losses through kiln surface. By applying these proposals they achieved 15.6% reduction in energy losses in the process. Pipilikaki et al. (2005) investigated the possibility of using tyre-derived fuel (TDF) as supplemental alternative fuel in cement plant. Quality measurements were carried out by using different qualitative analytical techniques such as X-ray diffraction (XRD), X-ray fluorescence (XRF) and optical microscopy in two series of raw mill, clinker and fuel samples with and without the use of TDF. In this specific study, TDF was used as the 6% of the total fuel. It was found that there was no apparent problem associated with utilizing TDF as a supplemental fuel in the clinker burning. Zabaniotou & Theofilou (2008) studied possibility of utilization municipal sewage sludge as a partial alternative fuel at a cement plant in Cyprus in order to consume unused wastes traditionally considered an environmental problem as renewable fuel that not only produces energy, but reduces pollutants emissions as well. In their work, environmental gaseous emission with emphasis on heavy metal concentrations, specially mercury (Hg) were measured. They reported that heavy metal concentrations in gas amounted to only 16% of the allowable levels and dioxin/furans amounted to only 6% of the allowable levels. It was also estimated that the saving out of using 7.5% wet sludge in this plant as supplementary fuel can reach to 8.0 euro/h. Conesa et al. (2008) analyzed the emission of different pollutants when partially substitute the typical fuel with tyre and two types of sewage sludge. In their experiments, different fuel samples containing sludge and/or tyres were tested and emission of Dioxins and furans, polycyclic aromatic hydrocarbons (PAHs) and other hydrocarbons, heavy metals, HCl, HF, CO, CO₂, NO_x and other parameters of the stack were analyzed for more than 1 year. It was found that the emission of PAHs and dioxins seems to increase with the amount of tyres fed to the kiln, probably due to the fed point used for this waste. Wang et al. (2009) conducted exergy analysis and optimization approach using genetic algorithm (GA) to find the most efficient case among different cogeneration systems including single flash steam cycle, dual-pressure steam cycle, organic Rankine cycle (ORC) and the Kalina cycle aimed to reuse waste heat from the exhaust gas and air vent streams in cement plant. The results of exergy analysis showed that thermal heat losses in turbine, condenser, and heat recovery vapor generator are relatively large and optimization strategy indicated that the Kalina cycle could achieve the best performance in cement plant. Sögüt et al. (2010) applied exergy analysis in a cement plant in Turkey. Analysis showed that 51% of the initial energy of the process was lost. Moreover a mathematical model was developed in order to assess the possibility of using heat losses to supply thermal energy for dwellings in the vicinity. They proved that by using thermal energy

losses instead of coal and natural gas it was possible to decrease domestic coal and natural gas consumption by 51.55% and 62.62% respectively and also to reduce CO₂ emissions by 5901.94 kg/hr and 1816.90 kg/hr. Kabir et al. (2010) conducted thermal energy audit analysis on the pyroprocessing unit of the cement plant, and found that flue gases and kiln shell lost 27.9% and 11.97% of initial thermal energy respectively. In order to enhance the energy performance and to reduce heat losses, (WHRSG) and Secondary kiln shell were considered. It was indicated that 5.30 MW of thermal energy losses and also 14.10% of Greenhouse Gases can be reduced by applying (WHRSG) and Secondary kiln shell. Rovira et al. (2011) evaluated human health risks for the population living around the cement plant in Spain. The objective of their study was to compare temporal trend of the environmental levels of polychlorinated dibenzo-p-dioxins and dibenzofurans (PCDD/Fs) and a number of metals (As, Cd, Co, Cr, Cu, Hg, Mn, Ni, Pb, Sn, Tl, V, and Zn). After 1 year utilizing sewage sludge and those obtained in previous studies performed when petroleum coke was exclusively used as source of required energy. Results showed an acceptable range according to international standards which support and encourage the option of using sewage sludge as fuel in processes without apparent additional health risks for the population living in the vicinity.

In this work, a cement plant is selected for study. The purpose of this study is to investigate the environmental and economic benefits of replacement of fuel oil with natural gas in DASHTESTAN cement plant, and also feasibility of using stack gas waste heat for steam generation. The process is briefly described first. Then, the process is simulated through commercial process simulator. After that, several scenarios will be proposed and discussed to make the process more efficient in terms of energy saving and pollutant reduction.

1. PROCESS DESCRIPTION

DASHTESTAN cement plant is located in the south of Iran, near the north-western Persian Gulf bank. It produces 150 ton/hr clinker and is consisted of three major stages. The first stage is mixing and crushing stage which is responsible to mix certain proportion of clay and limestone and to crush them to powder form. The second and the most important stage is pyroprocessing which consumes more than 90% of total energy (Kabir et al., 2010) and in consequence generates more than 90% of CO₂ emitted in the process. This stage includes preheater, calciner, kiln and cooler. Clinker as nodular and well mixed form is the main production of this stage which then is sent to final grinding and crushing stage. During third stage, clinker is crushed and grinded to tiny grinds in powder form, mixed with additives and then is sent to packaging stage. Schematic of DASHTESTAN cement plant is shown in Figure 1.

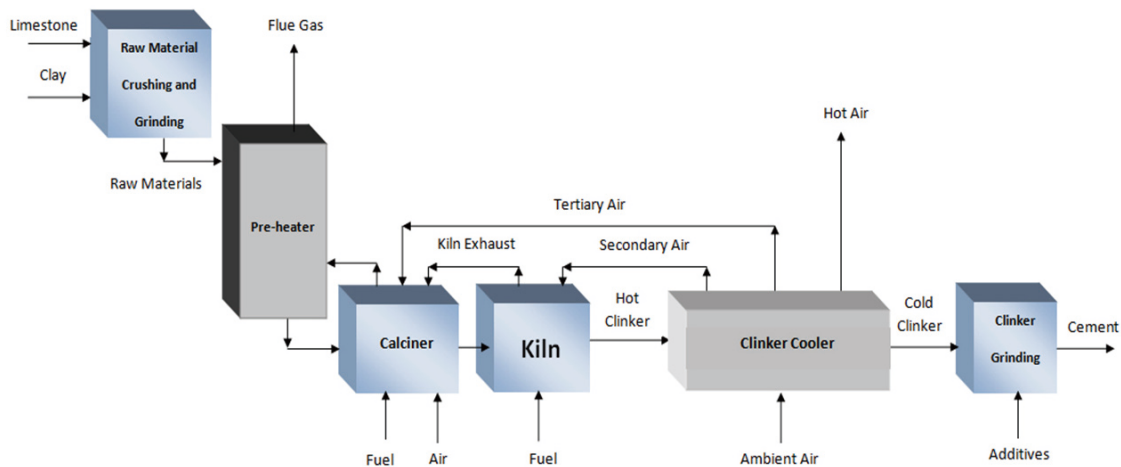


Figure 1
DASHTESTAN Cement Plant Block Diagram

As mentioned before, pyroprocessing is the most important stage in terms of energy and pollution management in cement plant. In this stage, raw materials mixture of clay and limestone with composition mentioned in Table 1 is crushed and mixed in the first crushing stage reaches to 50 μ m grain size. This solid mixture is entered to the preheater at 50°C. The preheater step is consisted of five cyclones which are used to preheat raw materials and separate them from the preheater gas. In this stage, feed stream with 50°C is heated up to 760°C by heat exchange with flue gases entering at 850°C from calciner. Flue gases then leave the preheater stage at 315°C and discharged to the environment. Such a high temperature is a valuable source of thermal heat. Solid stream at 760°C leaves preheater stage and is preceded to claciner where two reactions occur to convert 95% of CaCO₃ and all amount of MgCO₃ to CaO and MgO respectively. These endothermic reactions which are the major source of CO₂ production in cement plant require too much thermal energy. This thermal heat is provided through 3 different sources including combustion of fuel (fuel oil), hot gases produced in the kiln as well as hot air discharged from the coolers.

The main reactions of each section are given in Table 2, along with other data as operating temperature and heat of reaction (Mintus et al., 2006; Kaantee et al., 2004; Mujumdar & Ranade, 2006). The reaction products are solid materials which consist of remained CaCO₃, CaO,

MgO, and the other materials. This solid stream is entered to the kiln at 890°C, where C₂S, C₃S, C₃A and C₄AF are produced as four major components of clinker under several chemical and physical (phase change) reactions presented in Table 2. The required thermal heat is provided by combustion of fuel oil and hot air sent from coolers. The composition of produced clinker is given in Table 3. The clinker enters the cooler with 12 fans at 1500°C and cooled down to 80°C. Ambient air which is sent to hot clinker is separated to three streams and different temperatures by ducts. One stream at 980°C is sent to the kiln, another one at 900°C is launched to calciner and the last one at 230°C is discharged to the environment as the second major source of thermal heat in cement plant.

Table 1
Raw Material Composition in DASHTESTAN Cement Plant

Component	Composition (%wt)
CaO	41.51
MgO	2.59
Al ₂ O ₃	3.39
Fe ₂ O ₃	2.54
SiO ₂	14.03
SO ₃	0.30
K ₂ O	0.57
Na ₂ O	0.24
L.O.I	34.83

Table 2
Chemical and Physical Reactions in DASHTESTAN Cement Plant

Reaction name	Reaction	Temperature range (°C)	Heat of reaction (Δ HR)
Decalcination	$\text{CaCO}_3 \longrightarrow \text{CaO} + \text{CO}_2$	700 - 900	+179.4 kJ mol ⁻¹
MgCO ₃ dissociation	$\text{MgCO}_3 \longrightarrow \text{MgO} + \text{CO}_2$	700 - 900	+117.61 kJ mol ⁻¹
β -C ₂ S formation	$2\text{CaO} + \text{SiO}_2 \longrightarrow \beta\text{-C}_2\text{S}$	900 - 1200	-127.6 kJ mol ⁻¹
C ₃ S formation	$\beta\text{-C}_2\text{S} + \text{CaO} \longrightarrow \text{C}_3\text{S}$	1200 - 1280	+16 kJ mol ⁻¹
C ₃ A formation	$3\text{CaO} + \text{Al}_2\text{O}_3 \longrightarrow \text{C}_3\text{A}$	1200 - 1280	+21.8 kJ mol ⁻¹
C ₄ AF formation	$4\text{CaO} + \text{Al}_2\text{O}_3 + \text{Fe}_2\text{O}_3 \longrightarrow \text{C}_4\text{AF}$	1200 - 1280	-41.3 kJ mol ⁻¹
Liquid clinker formation	$\text{Clinkersol} \longrightarrow \text{Clinker}_{\text{liq}}$	>1280	+600 kJ kg ⁻¹ clinker

Table 3
Clinker Composition Produced in DASHTESTAN Cement Plant

Component	Composition (wt %)
C ₂ S	22.42
C ₃ S	52.62
C ₃ A	7.34
C ₄ AF	11.88
SO ₃	0.66
K ₂ O	0.87
Na ₂ O	0.47
MgO	3.74

2. PROCESS SIMULATION

Process simulation is the best approach in order to assess the possibility of a new proposed process or proposed changes in a current process. On the other hand, it is time and budget consuming to test a new process or any proposed changes experimentally in the laboratory or in the pilot plant. Therefore, in order to assess possibility of fuel substitution from fuel oil to natural gas in DASHTESTAN cement plant and evaluate the use of waste heat for steam generation, simulation scenarios were proposed. The simulation of process and scenarios were conducted in the following manner. First, the existing plant was simulated by HYSYS, version 3.0.1. To do this, details of raw materials and kinetic reactions information

were added to the component data bank. Also, the fuel was defined for simulator by adding different laboratory analysis reports. Then, the existing cement production process was simulated as base case. The simulation results were then validated by available plant operating data. Then, a scenario was defined to alter the plant fuel from fuel oil to natural gas. In this step, modifications were made to the plant to meet change of fuel, and analysis of stack gas was compared to the base case which applies fuel oil. Another scenario was proposed to use the waste heat of stack gas as a heat source to produce steam. The feasibility of altering fuel to natural gas and steam generation in conjunction with cement production is proposed as energy efficient, less pollutant process. The main assumptions are as follow:

1. Process is operated at atmospheric pressure and pressure drop is negligible;
2. Heat losses through pipes and heat exchangers are ignored;
3. Adiabatic conversion reactors have been used as calciner and the kiln;
4. Heat exchangers are used as clinker cooler and steam generator;
5. Preheater stage is a complex of five counter-current flow separators and mixers;

The composition and heating value of fuels used in this study are shown in Table 4 (Iran, Ripi, 2010)

Table 4
Properties of Fuels Used in Process Simulation

Fuel	C(%wt)	H(%wt)	N(%wt)	S(%wt)	Methane(%mol)	Ethane(%mol)	MW	HHV (MJ/kg)
Fuel Oil	85.25	11.8	0.2	2.75	-----	-----	211.6	43.5
Natural Gas	-----	-----	-----	-----	98.65	1.35	16.23	51

3. RESULTS AND DISCUSSION

The details of different scenarios to investigate change of fuel and reuse of waste heat for steam generation are summarized in Table 5.

Using data collected from control room and chemical laboratories, the base case process was simulated with its process flow diagram shown in Figure 2. The pyroprocessing stage of this cement plant consists of four main parts, including:

1. Preheater, a complex of 5 mixers and separators to preheat raw material and separate them from hot gas;

2. Calciner, which is a conversion reactor to calcinate 95% of CaCO₃ and also to decompose all amount of MgCO₃;

3. Kiln, which is a conversion reactor to provide a suitable space for all the reactions to produce nodular and well mixed clinker;

4. Coolers, which are heat exchangers to cool hot clinker down and prepare it for final grinding stage. The simulation results were used to estimate pollutants emission, heat losses and fresh air used in different parts of the process.

Table 5
Summary of Studied Scenarios

Case	Fuel type	Waste heat recovery	Additional facilities used
I	Fuel Oil	–	Base Case
II	Fuel Oil	+	1. Two boilers for steam generation
III	Natural Gas	–	1. One heat exchanger with 157,230 kg/hr air capacity
IV	Natural Gas	+	2. One fan with the same capacity 1. Two boilers for steam generation 2. Two pumps to increase water pressure from 1bar to 11.35 bar 3. One heat exchanger with 157,230 kg/hr fresh air capacity 4. One fan with the same capacity of fresh air

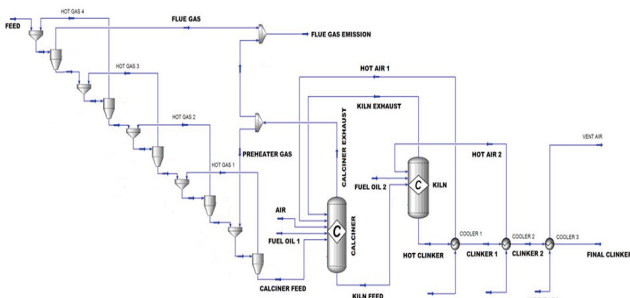


Figure 2
DASHTESTAN Cement Manufacturing Process Flow Sheet

3.1 Steam Generation Unit

As mentioned earlier vent air and flue gas streams are two significant sources of heat losses in cement plant. In order to recover heat from these two streams and reuse a part of energy, Waste Heat Recovery Steam Generator (WHRS) is proposed. This process is sometimes referred as economizer, as it can make use of waste heat more efficiently. Block diagram of this option is shown in Figure 3. The produced steam can be used directly in the process; alternatively, it may be used as a source for electricity generation through steam power plant.

Simulation results show that if temperature of exhaust streams from WHRS be set in 100°C, the vent air and flue gas streams will generate 7,528 and 25,114 kg/hr steam (180.7°C, 11.35 bar), respectively.

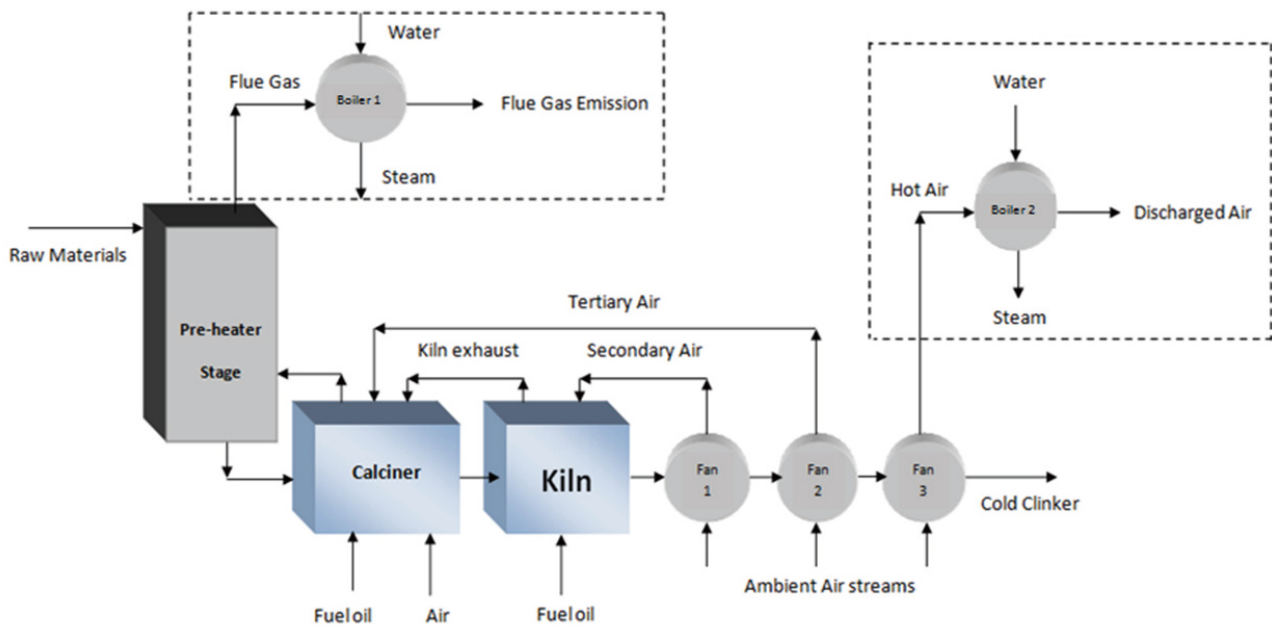


Figure 3
Block Diagram of Base Process Including Steam Generation

3.2 Utilizing Natural Gas as an Alternative Fuel and Steam Generation Unit

The objective of this scenario is to study the feasibility of using sweet natural gas as an abundant source of energy available in the area. Natural gas is a low-carbon fuel with high hydrogen-to-carbon (H/C) ratio, and has many advantages against fuel oil from environmental aspects, and produces less pollutant when burnt. The sweet natural gas contains neither sulfur nor nitrogen-containing compounds. Therefore, emission of SO_x and NO_x pollutants is diminished.

The PFD of plant utilizing natural gas as fuel is shown in Fig. 4. The process is similar to the case with fuel oil, except for an additional heat exchanger used to preheat combustion air sent to calciner chamber. Moreover simulation shows that from vent air and flue gas streams 5,326 and 44,156 kg/hr steam (180.7°C, 11.35 bar) can respectively be generated if temperature of exhaust streams from WHRSG be set in 100° C.

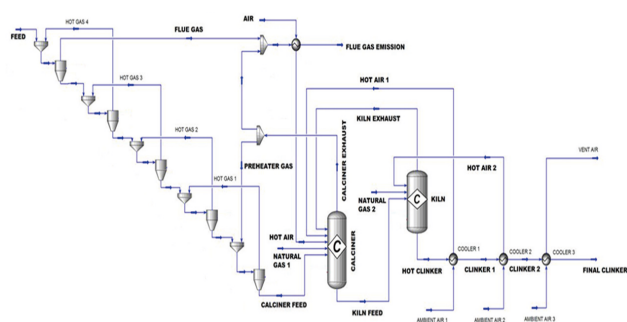


Figure 4
Proposed Process Flow Diagram Utilizing Natural Gas

Table 6 summarizes flow rate and temperature of the main streams in the both base and proposed process. Composition of pollutants, as well as rate of fuel consumption in processes is given in Table 7 and Fig.5. The heat loss through major parts of plants is also shown in Table 8.

Table 6
Flow Rate and Temperature of the Main Streams in both Base and Proposed Process

Stream	Base case process simulation		Natural gas process simulation	
	Flow rate (kg/hr)	Temperature (°C)	Flow rate (kg/hr)	Temperature (°C)
Feed	232,360	50	232,360	50
Calciner Feed	231,200	760	231,200	760
Fuel oil 1	7,500	120	-----	-----
Natural gas 1	-----	-----	7,727	50
Air	35,000	50	157,230	30
Hot air	-----	-----	157,230	500
Calciner exhaust	313,970	850	447,740	850
Kiln Feed	153,770	890	153,770	890
Natural gas 2	-----	-----	4,854	50
Fuel oil 2	5,300	120	-----	-----
Hot Clinker	150,000	1500	150,000	1500
Final Clinker	150,000	80	150,000	80
Total Ambient Air	341,000	26.8	307,571	26.8
Hot Air 1	80,000	900	110,210	900
Hot Air 2	105,000	980	86,551	980
Vent Air	156,610	230	110,810	230
Kiln exhaust	114,060	1,150	95,164	1,150
Preheater gas	233,600	850	223,500	850
Flue Gas emission	315,130	315	448,900	350

Table 7
Fuel Consumption and Pollutant Emission in Two Processes

Process	Item (Kg/hr)				
	CO ₂ emission	NO ₂ emission	SO ₂ emission	Dust	Fuel
Base case	118,693	39.39	329.94	1,160	12,800
Process utilizing natural gas	115,367	0	0	1,160	12,581

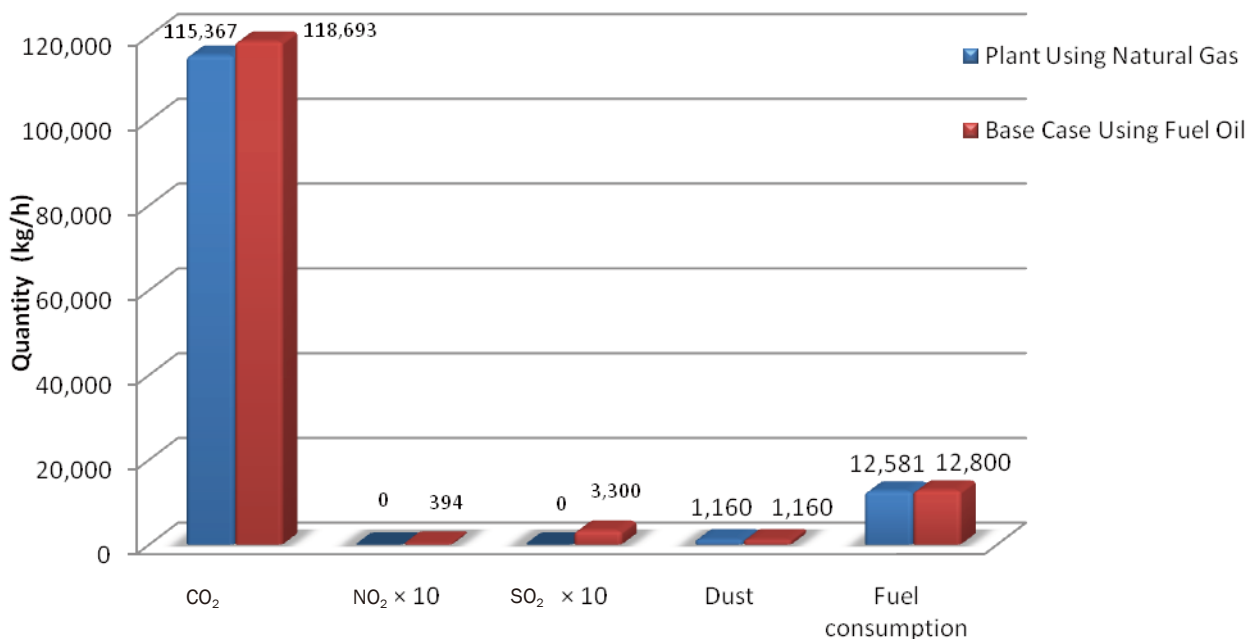


Figure 5
Comparison of Fuel Consumption and Pollutants Emission for Two Fuels

As Table 7 and Fig.5 clearly demonstrate using natural gas instead of fuel oil in this cement process can reduce 3,326 kg carbon dioxide per hour. Although this amount of reduction seems to be small quantity but when annual mitigation is taken to account, 26,342 ton carbon dioxide reduction will be resulted in a year which is significant approach to mitigate greenhouse gases emission and to subside impact of global warming.

Moreover since sweet natural gas contains neither sulfur nor nitrogen-containing compounds. Therefore,

emission of SO₂ and NO₂ pollutants is eliminated as compare to fuel oil.

As far as dust emission is concerned and by considering the nature of dust in cement process which is tiny particles of minerals carried by hot gases starting from coolers and emitting with flue gases, therefore and in result of the same quantity of raw materials and cement production in both processes, it can concluded that dust emission is independent of type of fuel and it is same in two processes.

Table 6
Flow Rate and Temperature of the Main Streams in both Base and Proposed Process

Major parts	Base case process		Natural gas process	
	Heat Loss (MJ/hr)	Losses of Initial Energy (%)	Heat Loss (MJ/hr)	Losses of Initial Energy (%)
Heat losses through vent air	21,180	3.82 %	15,000	2.32%
Heat losses through flue gas emission	70,660	12.71%	124,200	19.35%
Heat losses through the kiln	52,751	9.43%	50,078	7.82%
Heat losses through calciner and preheater	46,313	8.32%	47,661	7.43%
Total heat losses	190,904	34.28 %	236,939	36.92%

3.3 Economic Investigation

Although economic investigation of a process is one of the most complex, detailed and time consuming part of the research, it is essential to perform economic analysis to get a perspective whether or not a feasible engineering modification is economically attractive. Therefore, to

assure the economic advantages of proposed scenarios in addition to their environmental benefits, economic comparison between proposed cases and base case in term of fuel consumption, additional facilities used and producing steam has been done. This comprehensive investigation requires estimating capital cost, operating

cost as well as variable cost of the fuel, raw material, additives, salaries and so on. In this work, we have applied engineering cost estimation techniques (Peters et al., 2003) to prepare construction and production costs for each scenario. Marshall & Swift Equipment Cost Index (2010) was used in all cases. In short the purpose of this evaluation is to prove that in addition to environmental friendly features of the scenarios they are capable to bring remarkable economic benefits which encourage industries to implement them.

3.3.1 Plant Utilizing Fuel Oil Including Steam Generation

- Fixed extra costs

Since producing steam in this case requires some facilities such as two heat exchangers with 7,528 and 25,114 kg/hr steam generation capacity and two pumps to increase water pressure from 1bar to 11.35 bar with capacity of 7,528 and 25,114 kg/hr, these additional charges should also be considered. This additional charge to provide these facilities is about \$536,630.

- Income

As mentioned before, utilizing waste heat provided by vent air and flue gas streams 32,642 kg/hr steam at 180.7° C, 11.35 bar can be generated. The value of produced steam is 32,642 kg/hr × 4.5×10⁻³ \$/kg = \$146.89 /hr which annually comes to 146.89 \$/hr × 24 hr/day × 330 day/year = \$1,163,369 /year

Taking 5 years as the service life for new facilities and applying linear depreciation method, the net income for this scenario is estimated to be \$1,056,043/year.

3.3.2 Plant Utilizing Natural Gas

- Fixed extra costs

Since this plant needs facilities such as one heat exchanger with 157,230 kg/hr hot air production and a fan with the same capacity, their costs should also be included

in the estimation of fixed cost. In this way, by estimating \$4000 as the cost of air fan and \$1,013,040 for heat exchanger, the fixed cost is estimated to be \$1,017,040.

- Income

Natural gas price in 2010 in Iran was \$ 0.1/kg. Therefore, the net yearly cost of natural gas for this plant is estimated to be \$9,946,152 /year. On the other hand, 12,800 kg/hr fuel oil is consumed in base case, with its price estimated to be \$0.262/kg. Therefore, total yearly cost of fuel is estimated to be \$26,560,512/year. As a result, the net income from changing the fuel is \$16,614,360/year. Taking 5 years as the service life for new facilities and applying linear depreciation method, the net income for this scenario is estimated to be \$16,410,952/year.

3.3.3 Plant Utilizing Natural Gas Including Steam Generation

- Fixed extra costs

This case is similar to previous one except for adding two steam generators and two pumps to generate 49,482 kg/hr steam from flue gas emission and vent air streams. By considering two steam generators, two pumps, one heat exchanger and a fan, extra fixed cost is estimated to be \$2,105,940.

- Income

Steam generated in this case is 49,482 kg/hr which has an estimated value of \$1,763,538/year. Adding this value to the net income from changing the fuel, total income for this case is estimated as \$18,377,898 /year. Taking 5 years as the service life for new facilities and applying linear depreciation method, the net income for this scenario is estimated to be \$17,956,710/year.

Table 9 summarizes extra fixed costs, gross and net income, and payback period of all proposed cases as compared to base case. Calculations show that all modifications are attractive from economic point of view.

Table 9
Economic Comparison of Proposed Processes and Case Study

Scenario	Extra Fixed Cost (over existing plant), (\$)	Gross income, (\$/year)	Net Income (Subtracting Depreciation), (\$/year)	Payback Period, (months)
Adding waste heat recovery to Base Case	536,630	1,163,369	1,056,043	5.5
Changing fuel for Base Case without waste heat recovery	1,017,040	16,614,360	16,410,952	0.75
Adding waste heat recovery as well as changing fuel for Base Case	2,105,940	18,377,898	17,956,710	1.4

CONCLUSION

In this work, a cement plant located in the south west of Iran, northern bank of Persian Gulf was selected for analysis. The main objectives of the study were to find quantity and the sources of thermal heat loss in the

process, evaluate the feasibility of alternating fuel from fuel oil to natural gas and add a waste heat recovery for energy reuse.

It was shown that adding waste heat recovery systems will lower the temperature of stack gases from 315°C

and 350°C to 100°C in base case and process utilizing natural gas, respectively. This temperature gradient is capable to generate 32,642 and 49,482 kg/hr of steam in two processes. The best economic case in term of annual net income was estimated to be case III which is process utilizing natural gas including steam generation. The fixed extra costs in this process and payback period were estimated to be \$2,105,940 and 1.4 months, respectively.

As conclusion it is justified that alternating process fuel from fuel oil to natural gas is advantageous from both environmental and economical points of view.

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REFERENCES

- [1] Conesa, J.A., Galvez, A., Mateos, F., Martin-Gullon, I., & Font, R. (2008). Organic and Inorganic Pollutants from Cement Kiln Stack Feeding Alternative Fuels. *Journal of Hazardous Materials*, 158, 585-592.
- [2] Engin, T., & Ari, V. (2005). Energy Auditing and Recovery for Dry Type Cement Rotary Kiln Systems, a Case Study. *Energy Conversion and Management*, 46(4), 551-562.
- [3] Gabel, K., & Tillman, A.M. (2005). Simulating Operational Alternatives for Future Cement Production. *Cleaner Production*, 13, 1246-1257.
- [4] Hendriks, C.A., Worrell, E., Price, L., Martin, N., & Ozawa Media, L. (1999). The Reduction of Greenhouse Gas Emission from the Cement Industry. *IEA Greenhouse Gas R&D programme*.
- [5] Research institute of petroleum industry. (2010). Iran, Ripi.
- [6] Kaantee, U., Zevenhove, R., Backman, R., & Hupa, M. (2004). Cement Manufacturing Using Alternative Fuels and the Advantages of Process Modeling. *Fuel Processing Technology*, 85, 293-301.
- [7] Kabir, G., Abubakar, A.I., & El-Nafaty, U.A. (2010). Energy Audit and Conservation Opportunities for Pyroprocessing Unit of a Typical Dry Process Cement Plant. *Energy*, 35, 1237-1243.
- [8] Khurana, S., Banerjee, R., & Gaitonde, U. (2002). Energy Balance and Cogeneration for a Cement Plant. *Applied Thermal Engineering*, 22, 485-494.
- [9] Marshall & Swift Equipment Cost Index, Marshall & Swift. (2010). Retrieved from http://web.ist.utl.pt/~ist11061/de/IndPrecos/Che-2010_04_Econ-Indic.pdf
- [10] Mintus, F., Hamel, S., & Krumm, W. (2006). Wet Process Rotary Cement Kilns: Modeling and Simulation. *Clean Techn Environ Policy*, 8, 112-122.
- [11] Mokrzycki, E., Bochenczyk, A.U., & Sarna, M. (2003). Use of Alternative Fuels in the Polish Cement Industry. *Applied Energy*, 74, 101-111.
- [12] Mujumdar, K.S., & Ranade, V.V. (2006). Simulation of Rotary Cement Kilns Using a One-Dimensional Model. *Chemical Engineering Research and Design*, 84(A3), 165-177.
- [13] Nazmul, H. (2005). *Techno-Economic Study of CO₂ Capture Process for Cement Plants* (Master dissertation, University of Waterloo, Ontario, Canada). Retrieved from <http://etd.uwaterloo.ca/etd/s5hassan2005.pdf>
- [14] Peters, M.S., Timmerhaus, K.D., & West, R.E. (2003). *Plant Design and Economics for Chemical Engineers*. In M.S. Peters (5th Ed). New York: McGraw-Hill.
- [15] Pipilikaki, P., Katsioti, M., Papageorgiou, D., Fragoulis, D., & Chaniotakis, E. (2005). Use of Tire Derived Fuel in Clinker Burning. *Cement & Concrete Composites*, 27, 843-847.
- [16] Prisciandaro, M., Mazziotti, G., & Veglio, F. (2003). Effect of Burning Supplementary Waste Fuels on the Pollutant Emissions by Cement Plants: a Statistical Analysis of Process Data, Resources. *Conservation and Recycling*, 39, 161-184.
- [17] Rasul, M.G., Widiyanto, W., & Mohanty, B. (2005). Assessment of the Thermal Performance and Energy Conservation Opportunities of a Cement Industry in Indonesia. *Applied Thermal Engineering*, 25, 2950-2965.
- [18] Rovira, J., Mari, M., Nadal, M., Schuhmacher, M., & Domingo, J.L. (2011). Use of Sewage Sludge as Secondary Fuel in a Cement Plant: Human Health Risks. *Environment International*, 37, 105-111.
- [19] Sögüt, Z., Oktay, Z., & Karakoç, H. (2010). Mathematical Modeling of Heat Recovery from a Rotary Kiln. *Applied Thermal Engineering*, 30, 817-825.
- [20] U.S.A, Environmental Protection Agency. (2010). *Inventory of U.S. Greenhouse Gas Emissions and Sinks:1990-2008*.
- [21] Wang, J., Dai, Y., & Gao, L. (2009). Exergy Analyses and Parametric Optimizations for Different Cogeneration Power Plants in Cement Industry. *Applied Energy*, 86, 941-948.
- [22] Zabaniotou, A., & Theofilou, C. (2008). Green Energy at Cement Kiln in Cyprus-Use of Sewage Sludge as a Conventional Fuel Substitute. *Renewable and Sustainable Energy Reviews*, 12, 531-541.